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[ADVANCED CHEMISTRY--CHAPTER 21: NUCLEAR CHEMISTRY](#)

Note: Highlighted items are the concepts addressed on the AP Exam, other information given is to compliment the AP Chemistry--Chapter 21: Nuclear Chemistry subject and outline information presented in the text in Chapter 21. Example Problems 1. What product is formed when radium-226 undergoes alpha decay? What isotope undergoes alpha decay to form lead-208? 2. Write nuclear equations for the following...

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[AP CHEMISTRY--CHAPTER 12: CHEMICAL KINETICS](#)

AP Chemistry--Chapter 12: Chemical Kinetics Practice Problems 1) The chart below gives the results from four experiments. Given the following equation and these results, determine the orders for all three reactants, the overall reaction order, and the value of the rate constant. $\text{BrO}_3(\text{aq}) + 5\text{Br}(\text{aq}) + 6\text{H}^+(\text{aq}) \rightarrow 3\text{Br}_2(\text{l}) + 3\text{H}_2\text{O}(\text{l})$
 Initial Initial Measured Experiment concentration concentration c...

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[CHAPTER 13 ? EXERCISE PROBLEMS](#)

AP Chemistry--Chapter 13: Chemical Equilibrium Practice Problems 1. Write the equilibrium expression for the following reaction: $4\text{NH}_3(\text{g}) + 7\text{O}_2(\text{g}) \rightarrow 4\text{NO}_2(\text{g}) + 6\text{H}_2\text{O}(\text{g})$ 2. The following equilibrium concentrations were observed for this reaction at 127C: $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightleftharpoons 2\text{NH}_3(\text{g})$ $[\text{NH}_3] = 3.1 \times 10^{-2} \text{ mol L}$ $[\text{N}_2] = 8.5 \times 10^{-1} \text{ mol L}$ $[\text{H}_2] = 3.1 \times 10^{-3} \text{ mol L}$ a. Calculate the value of K at 127C for this reaction. ...

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[MATHCAD - CHAPTER 5 PRACTICE PR](#)

Example 5.2 from text 765 kV 3 phase 60 Hz 300 km transposed line has $z = 0.0165 + j 0.3306 \text{ } \Omega/\text{km}$ $S_y = 4.674 \times 10^{-4} \text{ km}^{-1}$ $\arg z = 177.14 \text{ deg}$ $12 \text{ } 1 \text{ } z = y = 1.547 \times 10^{-2} \text{ m}^{-1}$ $\arg z = 88.6 \text{ deg}$ $1 \text{ } z = y = 1.244 \times 10^{-2} \text{ m}^{-1}$ $\arg z = 2.86 \text{ deg}$ $z = 7.082 \times 10^{-2} \text{ } \Omega/\text{km}$ $\arg z = 266.12 \text{ deg}$ $z = 1.429 \text{ deg}$ Notes square root of complex number is square root of its magnitude at one ha...





















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