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[PHYSICS UNIT 01 CVPM PACKET 2012](#)

Physics! Unit I CVPM Name: Constant Velocity Model
The front of each model packet should serve as a storehouse for things you'll want to be able to quickly look up later. We will usually try to give you some direction on a useful way to organize this space (see the table below).
Physical Quantity Description Symbol
Units CVPM Model Summary
Save this space for the end of the unit when your teacher will...

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[UNIT 6 ASSIGNMENTS ANSWERS](#)

pg. 16 Selected Chemistry Assignment Answers
Unit 6: Thermochemistry, States of Matter and Intermolecular Forces
Chapter 10: Causes of Change
10.1: Energy Transfer (Practice on pg. 342)
1. 97 J 2. 1.3 mol 3. 2.2 10² J 4. 197 K
(Section Review on pg. 344)
9. 29.2 kJ 10. 14.6 kJ 11. 0.52 mol 12. 358 K
13. 295 K 14. 0.234 J (K g) 15. 0.864 J (K g) 18. 100 J (K mol)
19. 0.6 J (K g)
10.2: Using Enthalpy (Practice...

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[UNIT 5 ASSIGNMENTS ANSWERS](#)

pg. 20 Selected Honour Chemistry Assignment Answers
Unit 5: Chemical Kinetics and Equilibrium
Chapter 14: Chemical Kinetics (pg. 489 to 493)
27. Activation energy is the amount of energy needed to overcome for the products to form. All reactions have some kind of activation energy. The higher is the activation energy, the slower the reaction rate.
30. Methane and oxygen gas mixture does not react spon...

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[UNIT 4 ASSIGNMENTS ANSWERS](#)

pg. 16 Selected Honour Chemistry Assignment Answers
Unit 4: Thermochemistry and Nuclear Chemistry
Chapter 6: Energy Relationships in Chemical Reactions (pg. 198 to 202)
1. System: - any specific part of the universe that is of interest to us. Surrounding: - the rest of the universe outside the system. Open System: - a system that can exchange mass and energy (usually in the form of heat) with its sur...





















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