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Chemistry--Unit 1: Atomic Structure and the Periodic Table Practice Problemsl. Atoms1) 100,000,000 copper atoms would form a line 1 cm long. How long would a line formed by 1×10^7 copper atoms be? Express your answer in millimeters. 1×10^8 Cu atoms 1×10^7 Cu atoms ? ? ? 1 mm 10 mm ? mmll. Structure of the Atom2) A sulfur-32 atom contains 16 protons, 16 neutrons, and 16 electrons. What is the m...

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CHAPTER 07 ATOMIC STRUCTURE AND PERIODICITY NOTES (ANSWERS)

Chemistry AP Unit 2: Chemical Bonding and Organic Chemistry Unit 2: Chemical Bonding and Organic Chemistry Chapter 7: Atomic Structure and Periodicity 7.1: Electromagnetic Radiation Electromagnetic (EM) Radiation: - energy that travels at the speed of light in a form of perpendicular waves. This includes everything from cosmic rays to radio waves. Wavelength (?): - the length of a wave (from cres...

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ELEMENT AND ATOMIC STRUCTURE SECTION 1 SECT1PP DOC S W TSE P 1

85 Q 1A substance X melts at 1600oC Its does not conduct electricity in the solid or in the moltenstate X probably hasA an ionic structureB a giant covalent structureC a simple molecular structureD a metallic structure85 Q 3The table below shows the number of neutrons and electrons in the following five atoms ionsatoms ions number of neutrons number of electronsP 8 82Q 12 10R2- 10 10S 12 10T 8 9Wh...

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